



Hydrogen Sensors

Health & Safety Executive/Health & Safety Laboratory
Stefan Ledin

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- ✧ Introduction
- ✧ Sensor capability wish list
- ✧ Overview of hydrogen sensor types
 - Existing sensor types
 - New technologies
 - Problems with sensors
- ✧ When things go wrong – A cautionary tale
- ✧ HSL hydrogen sensors
 - Type and specifications
 - Why we choose these types
- ✧ Summary and conclusions
- ✧ Acknowledgements

☀ Where might one find H₂ sensors?

- In the hydrogen generation process from carbon-containing fuel reforming or electrolysis;
- In hydrogen storage and distribution, at production sites and filling stations; and
- In hydrogen fuel cell/combustion systems. These can be stationary, for example power production or uninterruptible power supply, or mobile, for example automotive.

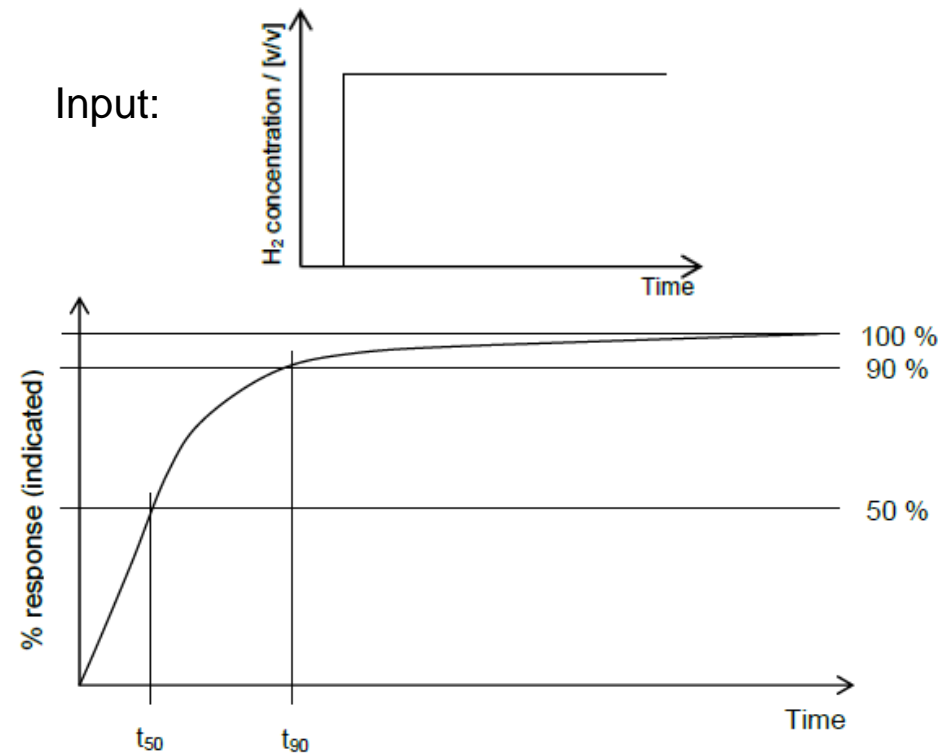
☀ Why do we use H₂ sensors?

- To detect leaks in the system
- To monitor hydrogen concentrations in experiments
- To monitor hydrogen concentrations in fuel cells

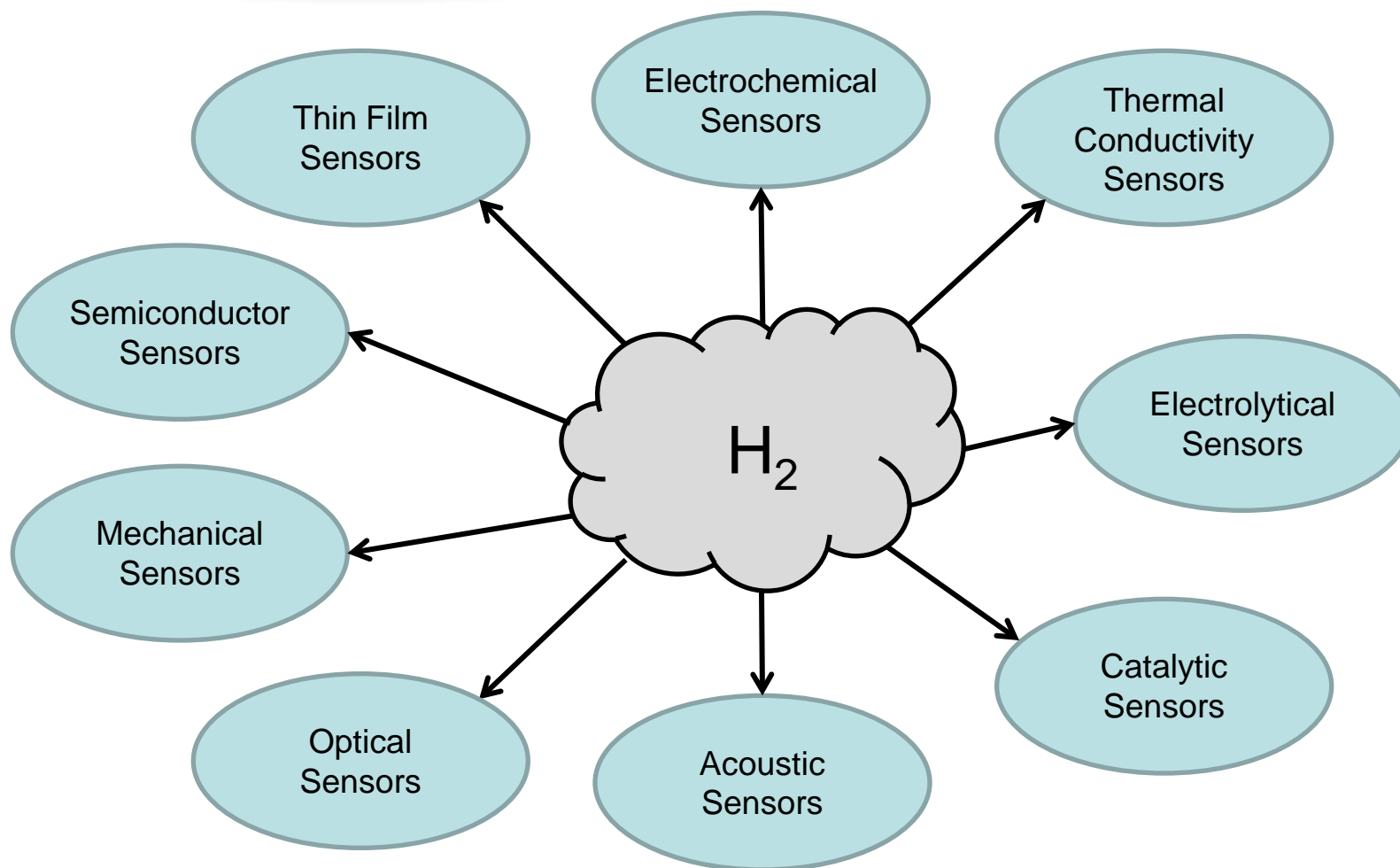
- ☼ Sensor range
 - 0-10 % v/v for safety
 - 0-100 % v/v for fuel cell systems
- ☼ Safe performance
- ☼ Reliable and accurate
 - Low uncertainty in the readings
- ☼ Stable signal with low noise
- ☼ Robustness
- ☼ Low sensitivity to
 - Temperature
 - Pressure
 - Relative humidity
 - Gas flow rates
- ☼ Fast response and recovery time
- ☼ Long life time
- ☼ Low cross sensitivity
- ☼ Low power consumption
- ☼ Low cost
- ☼ Small size
- ☼ Simple installation and maintenance
 - Long service intervals
- ☼ Simple integration into the system

We are not asking a lot of the sensors then ...

- There are a number of different types of hydrogen sensors
 - The method of detecting the gas differs
 - Their range of applicability
 - Their sensitivity to environmental conditions
 - Temperature
 - Pressure
 - Relative humidity
 - Other gases
 - Their response time
 - Usually expressed in terms of t_{90}
 - Their accuracy / resolution
 - Their susceptibility to poisoning



The response time t_{90} is defined as the time it takes for the output from the sensor to reach 90 % of its final value after being subjected to a step change in concentration



<i>Type</i>	<i>Principle</i>	<i>Advantages</i>	<i>Disadvantages</i>
Electrochemical	H ₂ oxidation at Platinum sensing electrode	<ul style="list-style-type: none"> • Quite selective • Low power consumption • Poison resistant 	<ul style="list-style-type: none"> • Some cross-sensitivity to CO • Narrow temperature range • Short lifespan of 2 years
Thin film	Reversible resistance increase	<ul style="list-style-type: none"> • Rapid response • Wide detection range • Does not need O₂ 	<ul style="list-style-type: none"> • Prone to poisoning • Sensitive to total pressure • Requires heating to 150 °C

<i>Type</i>	<i>Principle</i>	<i>Advantages</i>	<i>Disadvantages</i>
ChemFET	Hydrogen absorption	<ul style="list-style-type: none"> • Wide detection range • Does not need O₂ • Low power consumption 	<ul style="list-style-type: none"> • Prone to poisoning • Sensitive to total pressure • Needs to be heated to 150 °C
Catalytic	Catalytic oxidation on heated bead	<ul style="list-style-type: none"> • Acceptable lifespan(?) • Wide operational temperature range 	<ul style="list-style-type: none"> • Not selective • High power consumption • Poisoning • Requires 5-10 % v/v O₂ • High maintenance

<i>Type</i>	<i>Principle</i>	<i>Advantages</i>	<i>Disadvantages</i>
Thermal conductivity	High thermal conductivity changes	<ul style="list-style-type: none"> • Quite selective • Poison resistant • Long term stability • Does not require O₂ 	<ul style="list-style-type: none"> • Not as sensitive as ChemFETs • Cross-sensitive to helium • Requires heating
Semiconductor	Surface conductivity changes	<ul style="list-style-type: none"> • Commercially available • Acceptable lifespan 	<ul style="list-style-type: none"> • Not selective • High power consumption • Sensitive to humidity and temperature

<i>Type</i>	<i>Principle</i>	<i>Advantages</i>	<i>Disadvantages</i>
Mass Spectrometry	Charge change detector	<ul style="list-style-type: none"> • Specific • Low limit of detection 	<ul style="list-style-type: none"> • Expensive • Bulky • Fragile • Needs skilled operator
Ultrasonic	Detection of ultra sound	<ul style="list-style-type: none"> • Not susceptible to poisoning or humidity • Non-directional 	<ul style="list-style-type: none"> • Not specific to H₂ • Interference from background noise • Only detects high pressure leaks

Type	Range¹	Resolution¹	Response [s]	Lifespan [years]
Electrochemical	0.0-0.2 % 0.0-2.0 %	2-10 ppm	30	1-2
Thin film + ChemFET	0.1-100 % 10-1000 ppm (with ChemFET)	0.1 % 10 ppm (with ChemFET)	5	?
Catalytic	0-100 % LEL	1 % LEL	20	3
Thermal conductivity	0.0-10 %	0.5 % range	20	10+
Semiconductor	50-5000 ppm	50 ppm	30	?

Note: the lifespan of a sensor depends on a number of factors, for example where it is mounted, the environmental conditions in which it operates, maintenance regime, ...

1. Beware of change of units

<i>Type</i>	<i>Principle</i>	<i>Advantages</i>	<i>Disadvantages</i>
Thick film	Resistance change	<ul style="list-style-type: none">• Low cost• Simple method	<ul style="list-style-type: none">• Susceptible to poisoning
Optoelectronic	<ul style="list-style-type: none">• Induced mechanical stress• Interferometry• Optical characteristics	<ul style="list-style-type: none">• Very low power consumption• No electromagnetic interference• Intrinsically safe	<ul style="list-style-type: none">• Sensitive to humidity and temperature
Nanotechnology	Electrical resistance	<ul style="list-style-type: none">• Unproven	<ul style="list-style-type: none">• Unproven

<i>Type</i>	<i>Principle</i>	<i>Advantages</i>	<i>Disadvantages</i>
MOS Schottky diodes	Change in electronic properties	<ul style="list-style-type: none">• Low concentrations detected• Can operate in inert and oxygen-containing atmospheres	<ul style="list-style-type: none">• Response affected by presence of O₂
Surface acoustic wave	Sorption ⇒ change of resonance frequency	<ul style="list-style-type: none">• Low power• Micro-scale	<ul style="list-style-type: none">• Susceptible to water vapour

This is by no means an exhaustive list

- Introduction of mobile fuel cell applications (for example cars) in particular is driving the sensor technique development forward
- Techniques under development (in reduced order of research activity)
 - Electrical resistance based technologies (40 %)
 - Optical sensors (15 %)
 - Catalytic sensors (5 %)
 - Electrochemical sensors (5 %)
 - Acoustic sensors

Reference:

Hübert, T., Boon-Brett, L., Black, L., and Banach, U. (2011). Hydrogen Sensors-A review, *Sensors and Actuators B* **157**(2):329-352.

- There are a number of operational issues
 - Accuracy and range
 - Calibration—frequent and/or tricky
 - Drifting
 - Cross-sensitivity to other gases
 - Poisoning by other gases
 - Interference by background noise
 - Longevity
 - Response time
 - Sensitivity to pressure, temperature and/or relative humidity
 - False alarms (hydrogen leak detection) / interaction with the system
 - Optimal/appropriate siting
- The sensor type to use is likely
 - To be a compromise; or
 - To be a combination of different types of sensors

Case Study – Hydrogen deflagration in a machine

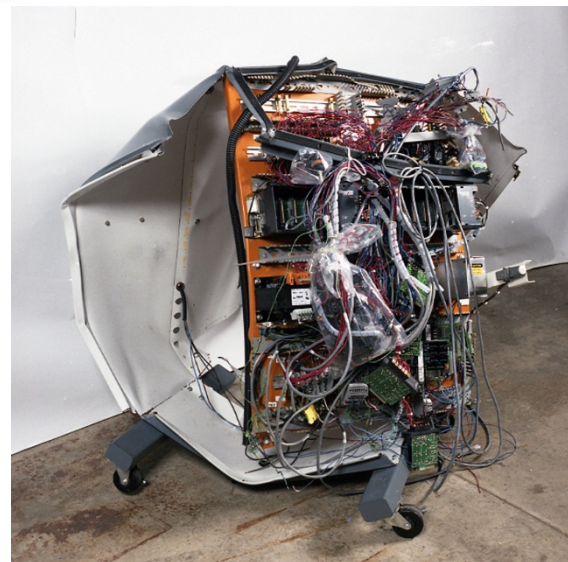
- ✦ In the machine, mechanical parts had metal vapour deposited onto its surface prior to being re-machined
- ✦ Hydrogen was being used in the process
- ✦ There was a build-up of hydrogen which was detected by the sensor
- ✦ However, the system did not shut down
- ✦ The hydrogen was ignited, resulting in a deflagration that injured the operator and caused damage to the machine

Lesson: Use hydrogen sensors to detect build-up of hydrogen, but also ensure that an “alarm” is acted upon.

When Things Go Wrong—2(2)



Gas flow meter cabinet—front view



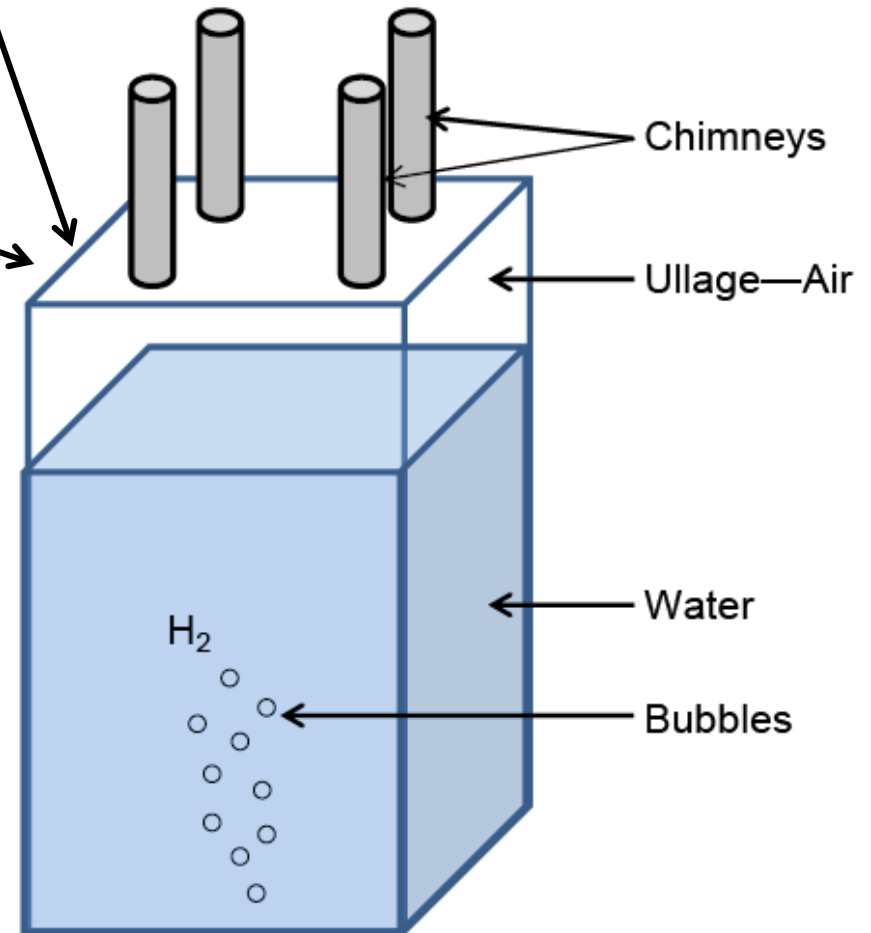
Machine – rear view



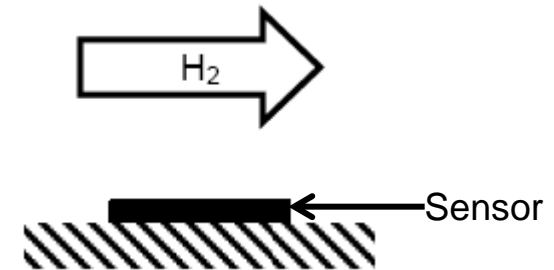
Flow meter cabinet – rear view

Types of Hydrogen Sensors @ HSL

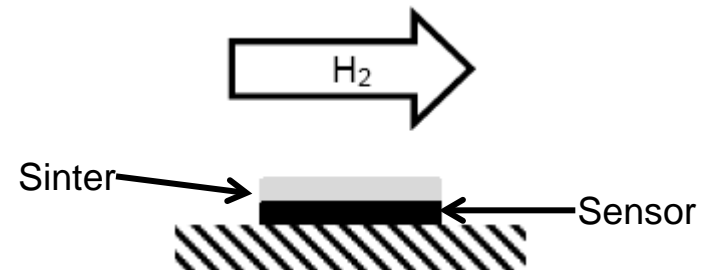
- XENSOR TCG-3880
 - Thermal conductivity sensor
- SEC H₂ 20000 4SE 5V
 - Electrochemical sensor
- VIAMED R17
 - Oxygen depletion sensor



- ✪ Thermal conductivity sensor
- ✪ Low voltage required: 8-45 mV
- ✪ Range: 0-100 % H₂
 - ➔ Realistic lower detection limit around 700-1000 ppm
- ✪ Readings may be affected by temperature and relative humidity
- ✪ Damp conditions can lead to corrosion of leads
- ✪ H₂ specific
- ✪ Response time:
 - ➔ 10 ms —naked sensor
 - ➔ 1000 ms — shielded sensor

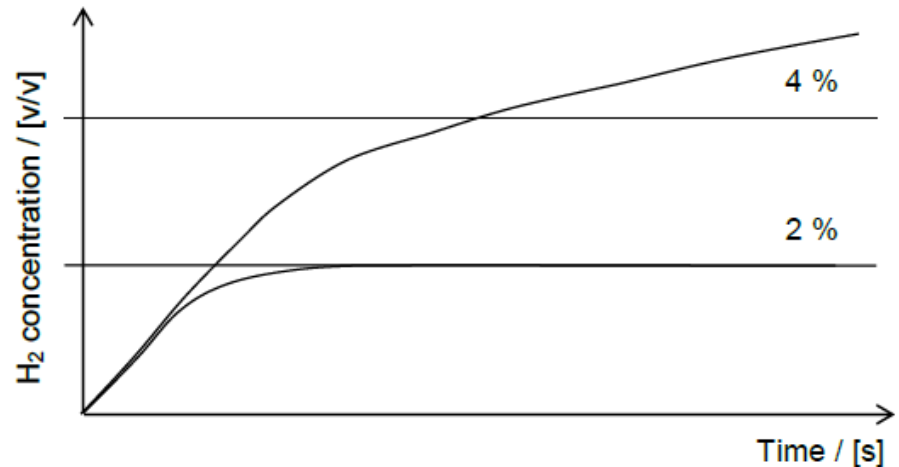


Naked sensor—reading affected by the flow of gas



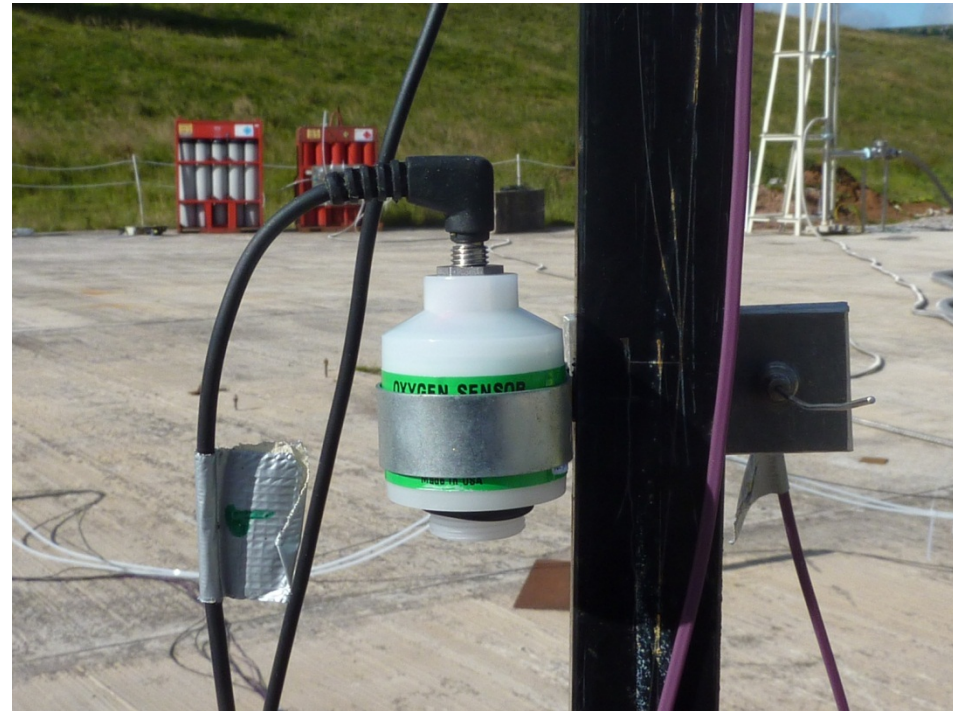
Shielded sensor

- ✪ Electrochemical sensor
- ✪ Hydrogen specific
- ✪ Solid State electrolyte
- ✪ Cross-sensitive
- ✪ Response time: 10-20 s
- ✪ Limited lifespan
 - ➔ 1-2 years (though the manufacturer claims 5 years)
 - ➔ Electrolyte is consumed
- ✪ Range:
 - ➔ 0-2 % v/v H₂ (std)
 - ➔ 0-4 % v/v H₂ (max)



- ✪ Sensor reading keeps drifting when using the 0-4 % v/v range ⇒
 - ➔ Calibration is tricky
 - ➔ What is the actual hydrogen concentration?

- ✧ Oxygen depletion sensor
- ✧ Insulated body, exposed head
- ✧ Silver foil to reduce effect of solar radiation
- ✧ Rapid response: t_{90} of the order of 2 s
- ✧ Range: 5-100 % v/v H_2
- ✧ Sensitive to
 - ➔ Temperature, pressure and relative humidity
 - Poor performance at low ambient temperatures ($< 5\text{ }^{\circ}\text{C}$) due to the compensation algorithm
- ✧ Requires frequent calibration
 - ➔ 10-20 min runtime max



Sensor in-situ

- VIAMED – Oxygen depletion sensor
 - Partly for historical reasons
 - This was almost the only available type of sensor at the time
- XENSOR TCG-5830 – Thermal conductivity sensor
 - Low voltage requirement
 - Fast response time
- SEC H₂ 2000 4SE 5V – Electrochemical sensor
 - Hydrogen specific
- Experience from the Instrumentation Special Interest Group in the HySafe Network of Excellence (EU FP6)
- XENSOR and SEC sensors are used together and appear to complement each other well

Note: A sensor of oxygen depletion type is not necessarily the most suitable choice of sensor for hydrogen measurements

Summary

- There are a number of different types of sensors
- Sensor choice-application specific
- Each type has its merits
- HSL has gained extensive experience of using:
 - Thermal conductivity sensors
 - Electrochemical sensors
 - Oxygen depletion sensors
- What we experienced when using the sensors (sensor type dependent!)
 - Rapid response times
 - Drifting
 - Constant recalibrations needed
 - Sensitivity to temperature, pressure and humidity

Conclusions

- These sensors are mostly fit for purpose, but
 - Calibration is vitally important
 - Drifting is an issue
 - Sensitivity to temperature, pressure and humidity
 - Poisoning or cross-sensitivity
- Oxygen depletion sensors are perhaps not the best option for hydrogen concentration measurements
- A combination of thermal conductivity and electrochemical sensors appears to work well
- A number of “new” technologies are coming on-stream

Colleagues at HSL
Colleagues at HSE
Partners in EU projects
Fuel Cell & Hydrogen/Joint Undertaking
European Commission

Thank You for Your Attention!

Any Questions?